

chapter 9 assessment chemistry pdf

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Test bank chapter (9) Choose the most correct answer 1. The two types of chemical bonds commonly found in compounds are: a) ionic and covalent. b) ionic and electrolytic. c) ionic and covalent. d) electrolytic and compound. 2. The electrons used by atoms to form chemical bonds are the: a) core electrons. b) valence electrons. c) lone pair electrons.

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Chapter 9 Practice Test - Naming and Writing Chemical Formulas Matching Match each item with the correct statement below. Match each item with the correct statement below. a. monatomic ion f. cation b. acid g. binary compound c. base h. anion d. law of definite proportions i. polyatomic ion e. law of multiple proportions ____ 1.

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Chapter 9 Assessment Chemistry Answers Chapter 9 Practice Test - Naming and Writing Chemical Formulas Matching Match each item with the correct statement below. Match each item with the correct statement below. a. monatomic ion f. cation b. acid g.

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Solutions Manual Chemistry: Matter and Change Chapter 9 141 Section 9.1 Reactions and Equations pages 282-288 Practice Problems pages 284-287 Write skeleton equations for the following word ... Section 9.1 Assessment page 288 7. Explain why it is important that a chemical

Chemical Reactions

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284 Chapter 9 Chemical Reactions Figure 9.3 Science, like all other disciplines, has a specialized language that allows specific information to be communicated in a uniform manner. This reaction between aluminum and bromine can be described by a word equation, a skeleton equation, or a balanced chemical equation.

Chapter 9: Chemical Reactions

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e. molar mass of $\text{Fe}(\text{NO}_3)_3 = 241.9 \text{ g}$ 272 g $\text{Fe}(\text{NO}_3)_3$ 1 mol $241.9 \text{ g} = 1.12 \text{ mol}$ $\text{Fe}(\text{NO}_3)_3$ f. molar mass of $\text{FeSO}_4 = 151.9 \text{ g}$ 12.7 mg FeSO_4 1 g 1000 mg 1 mol $151.9 \text{ g} = 8.36 \times 10^{-5} \text{ mol}$ FeSO_4 4 g. molar mass of $\text{LiOH} = 23.95 \text{ g}$ 6.91 g 103 g LiOH 1 mol $23.95 \text{ g} = 288.5 = 289 \text{ mol}$ LiOH 14. a. molar mass of $\text{CaCO}_3 = 100.1 \text{ g}$ $2.21 \times 10^{-4} \text{ mol}$ CaCO_3 100.1 g 1 mol = 0 ...

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CHEM 210 [CHAPTER 10: REACTIONS AND SYNTHESIS

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